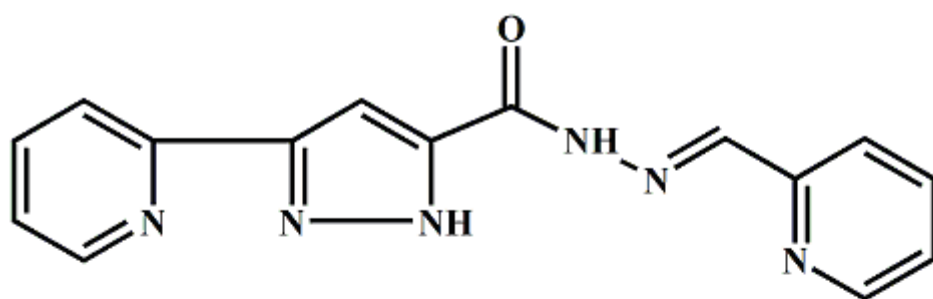
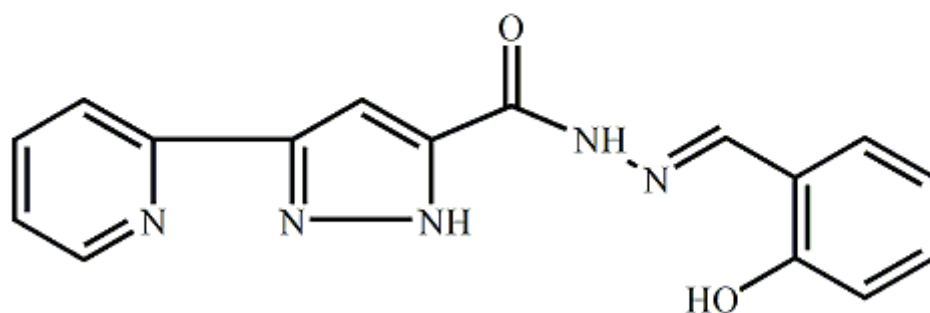


Construction of Polynuclear Lanthanide (Ln= Dy^{III}, Tb^{III} and Nd^{III}) Cage Complexes using Pyridine-Pyrazole based ligands: Versatile Molecular Topologies and SMM behavior

Sukhen Bala,^[a] Mousumi Sen Bishwas^[b], Bhaskar Pramanik^[c], Sumit Khanra,^[c] Katharina M. Fromm^[d], Pankaj Poddar^[b] and Raju Mondal^{[a]}*



3-(pyridin-2-yl)-*N'*-((pyridin-2-yl)methylene)-1*H*-pyrazole-5-carbohydrazide(H_2L_1)



***N'*-(2-hydroxybenzylidene)-3-(pyridin-2-yl)-1*H*-pyrazole-5-carbohydrazide(H_3L_2)**

Figure S1: Schematic diagram of ligands.

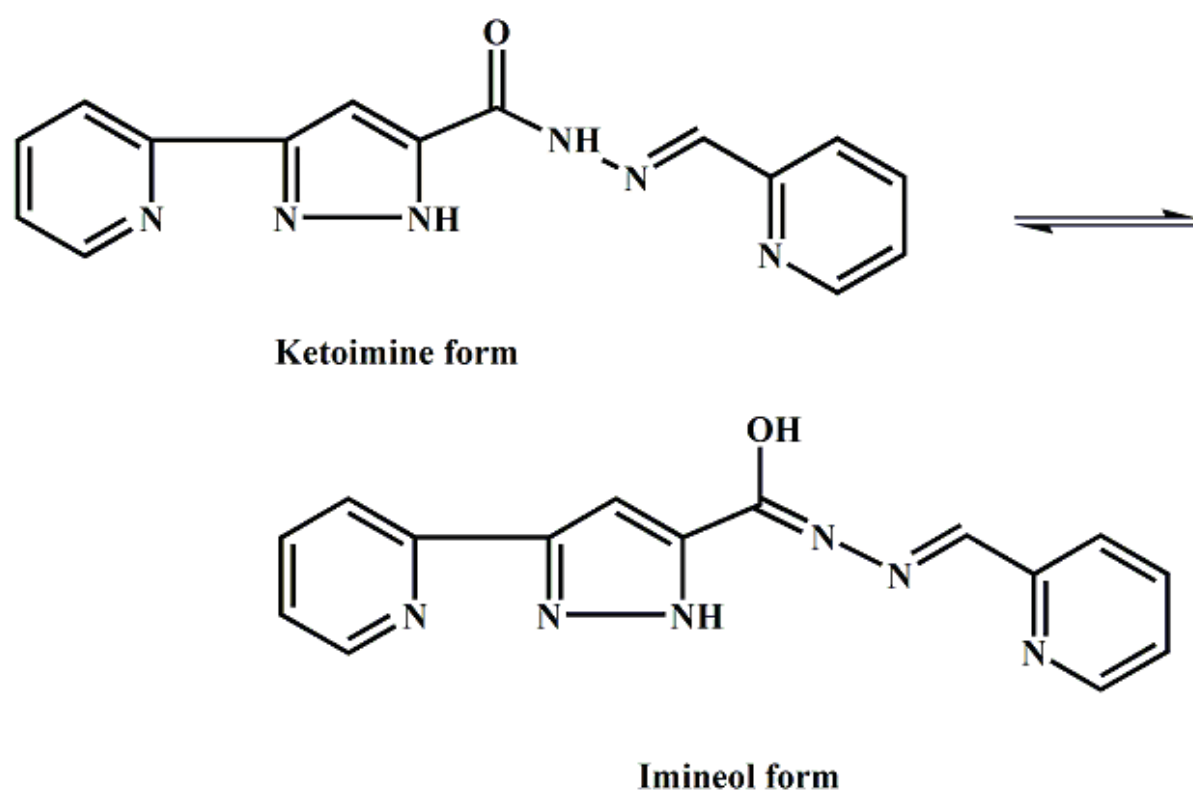


Figure S2: Keto–enol tautomerism of H_2L_1 .

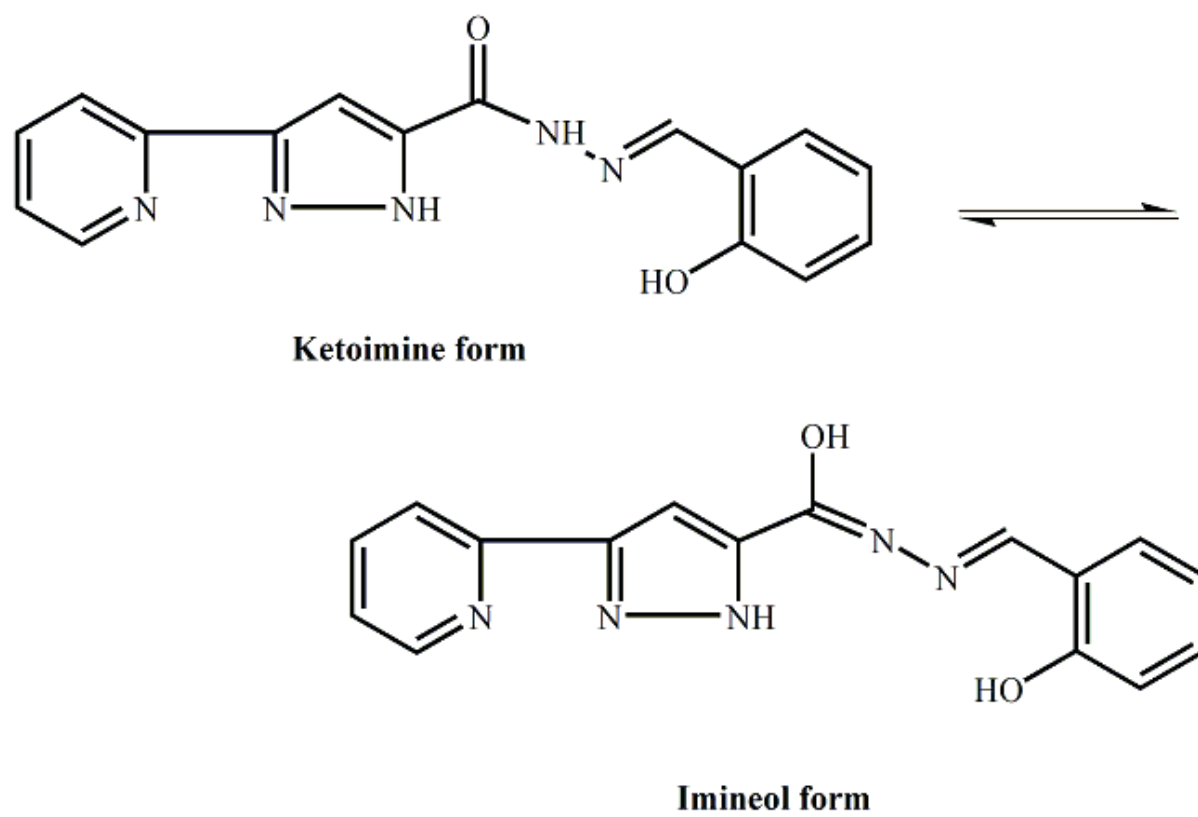


Figure S3: Keto–enol tautomerism of H_3L_2 .

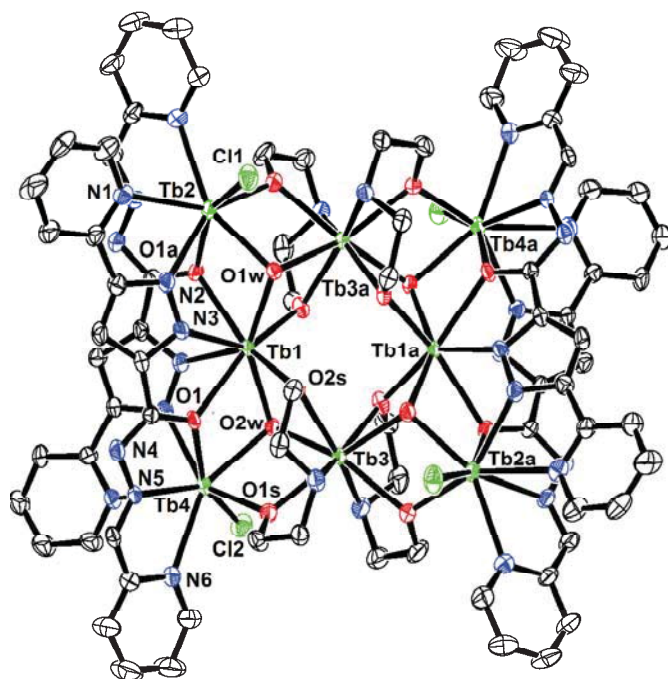


Figure S4. Perspective view of **2** with thermal ellipsoids at the 50% probability level. Hydrogen atoms are omitted for clarity.

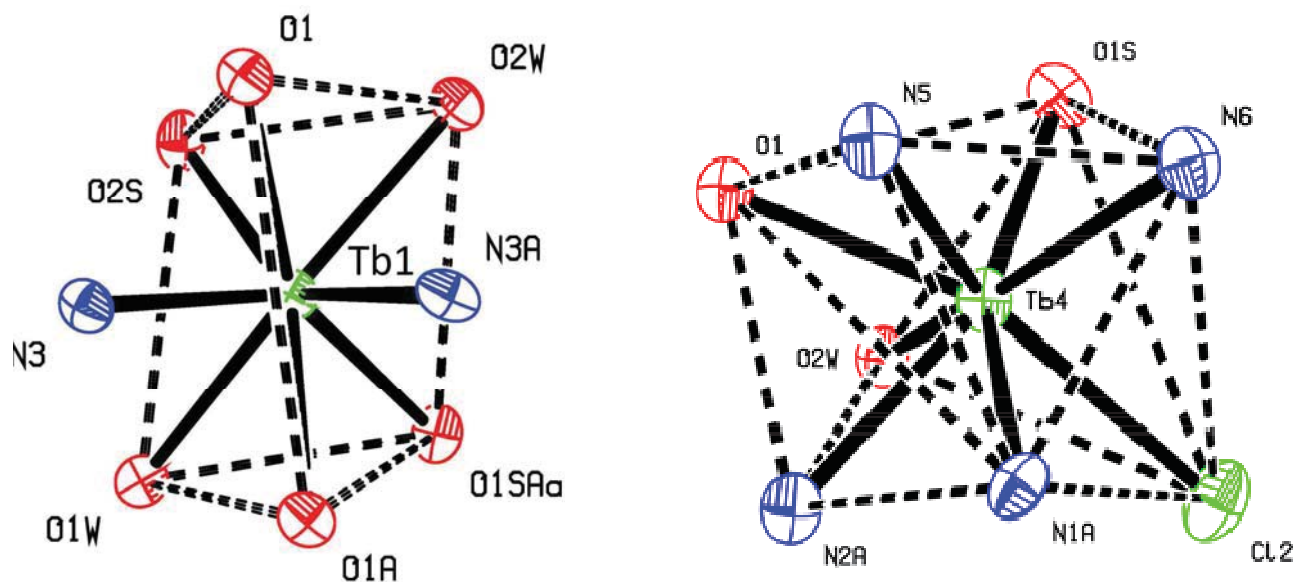
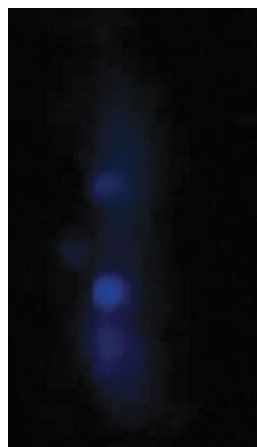
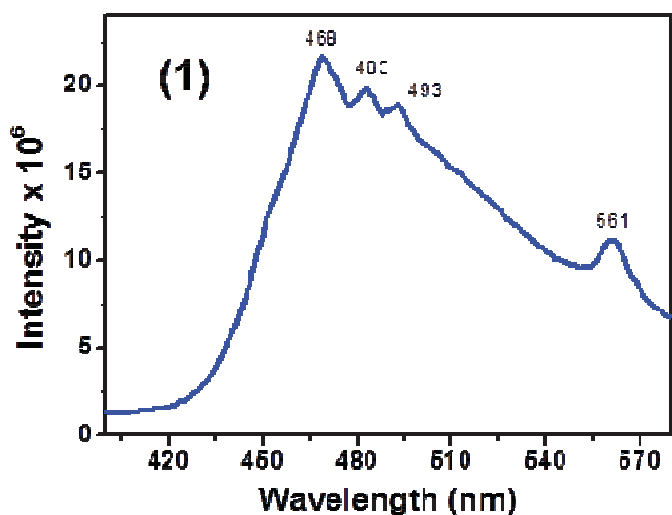


Figure S5. Representations of bicapped trigonal prism and distorted square antiprismatic geometry around Tb in **2**.

Luminescence Study:

Dy^{III} complexes **1** and **3** show very similar characteristic luminescence properties^{S1}. Upon excitation at $\lambda_{\text{ex}} = 265$ nm, a strong luminescence is attributed to complexes **1** and **3**. Both **1** and **3** exhibit typical blue emission peak at ~ 480 nm and another yellow emission peak at ~ 570 nm, characteristic for Dy^{III}.^{S1} The blue emission is assigned to a $^4F_{9/2} \rightarrow ^6H_{15/2}$ transition whereas the yellow emission is assigned to a $^4F_{9/2} \rightarrow ^6H_{13/2}$ transition as shown in Figure S6. The spectra of **1** and **3** shown in Figures S6 clearly indicate that the intensity of the blue emission is much higher than that of the yellow emission. Furthermore, the complexes display additional peaks at ~ 482 and ~ 493 nm.



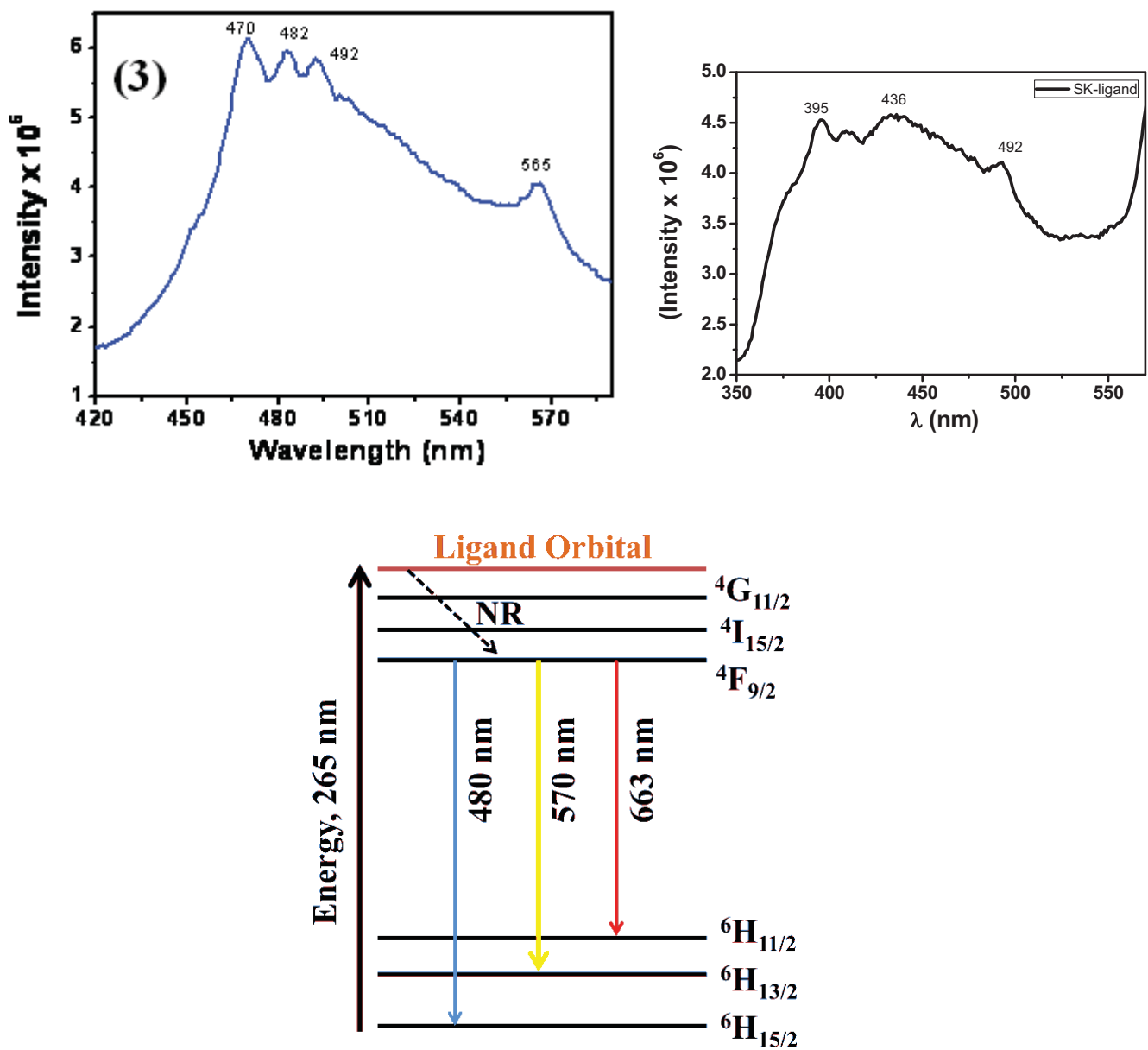


Figure S6. Solid state emission spectra of 1, 3 and ligand at 298 K, excited at 265 nm and energy level diagrams of Dy^{III} showing possible transitions.

Magnetic data points below 50 K:

The $\chi_m T$ -T for all the three compounds **1**, **2**, and **3** were measured again below 50 K with more precision. Figure S7, S8, S9 respectively shows the $\chi_m T$ -T plot for samples **1**, **2**, and **3** respectively. It was observed that the very small kink or anomaly that was seen ~ 27 K for compound **1** and **3** in early measurement was not visible when magnetization data were re-measured with more data points below 50 K.

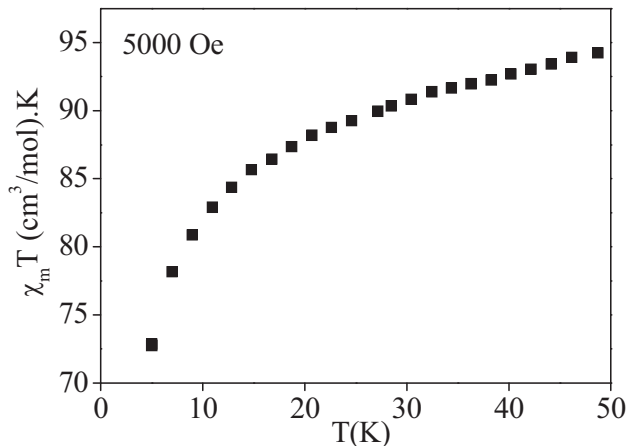


Figure S7. The $\chi_m T$ - T plot for the complex **1** under an applied field of 5000 Oe from 5 -50 K.

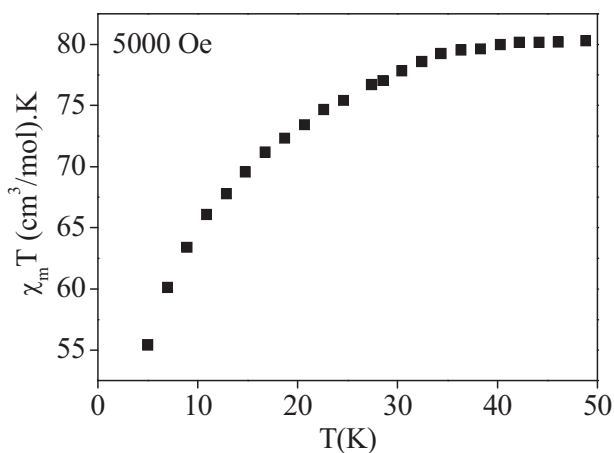


Figure S8. The $\chi_m T$ - T plot for the complex **2** under an applied field of 5000 Oe from 5 -50 K.

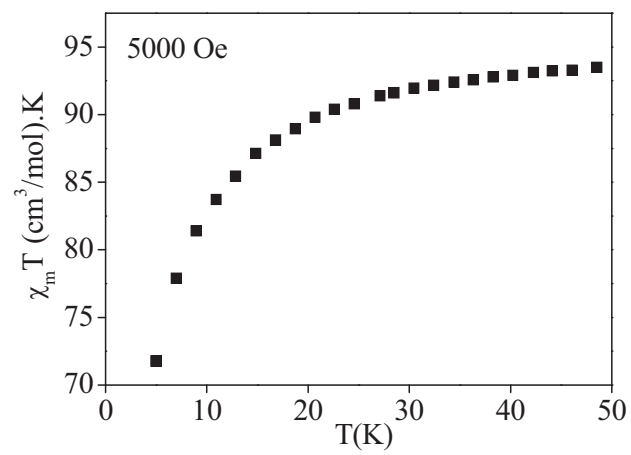


Figure S9. The $\chi_m T$ - T plot for the complex **3** under an applied field of 5000 Oe from 5 -50 K.

Frequency-dependent in-phase and out-of phase components of ac susceptibility of compound 1, 2 and 3:

The frequency dependence of compound 1, 2, and 3 from frequency 10 Hz to 1000 Hz are measured keeping temperature constant from 2 K to 20 K at an interval of 2 K. **Figures S10-S12**, show frequency dependence of in phase and out of phase component of susceptibility under zero dc field and 0.5 T dc field for compound 1-3.

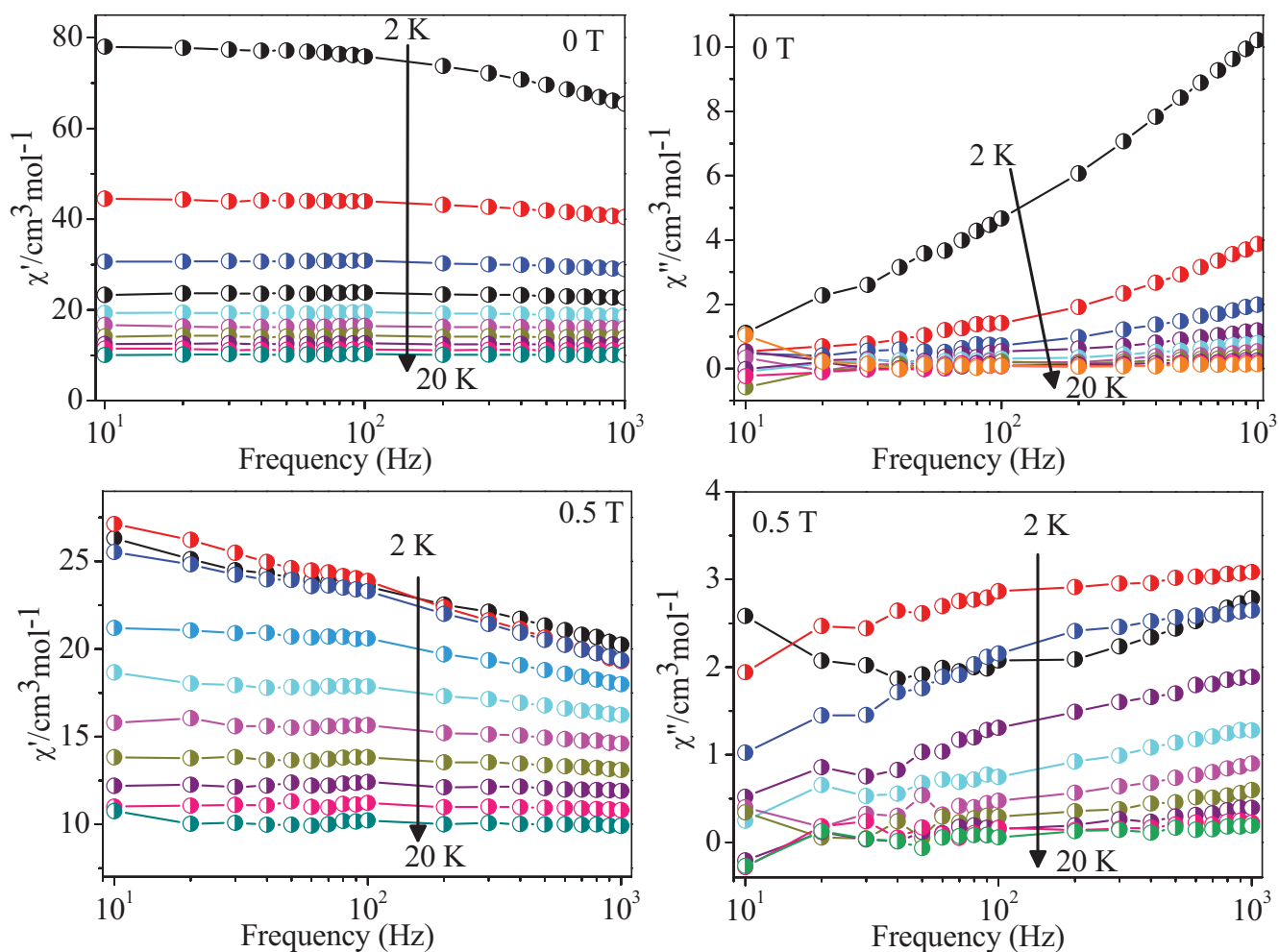


Figure S10: Frequency-dependent in-phase and out-of phase components of ac susceptibility of compound 1 under zero dc field (top) and 0.5 T field (bottom).

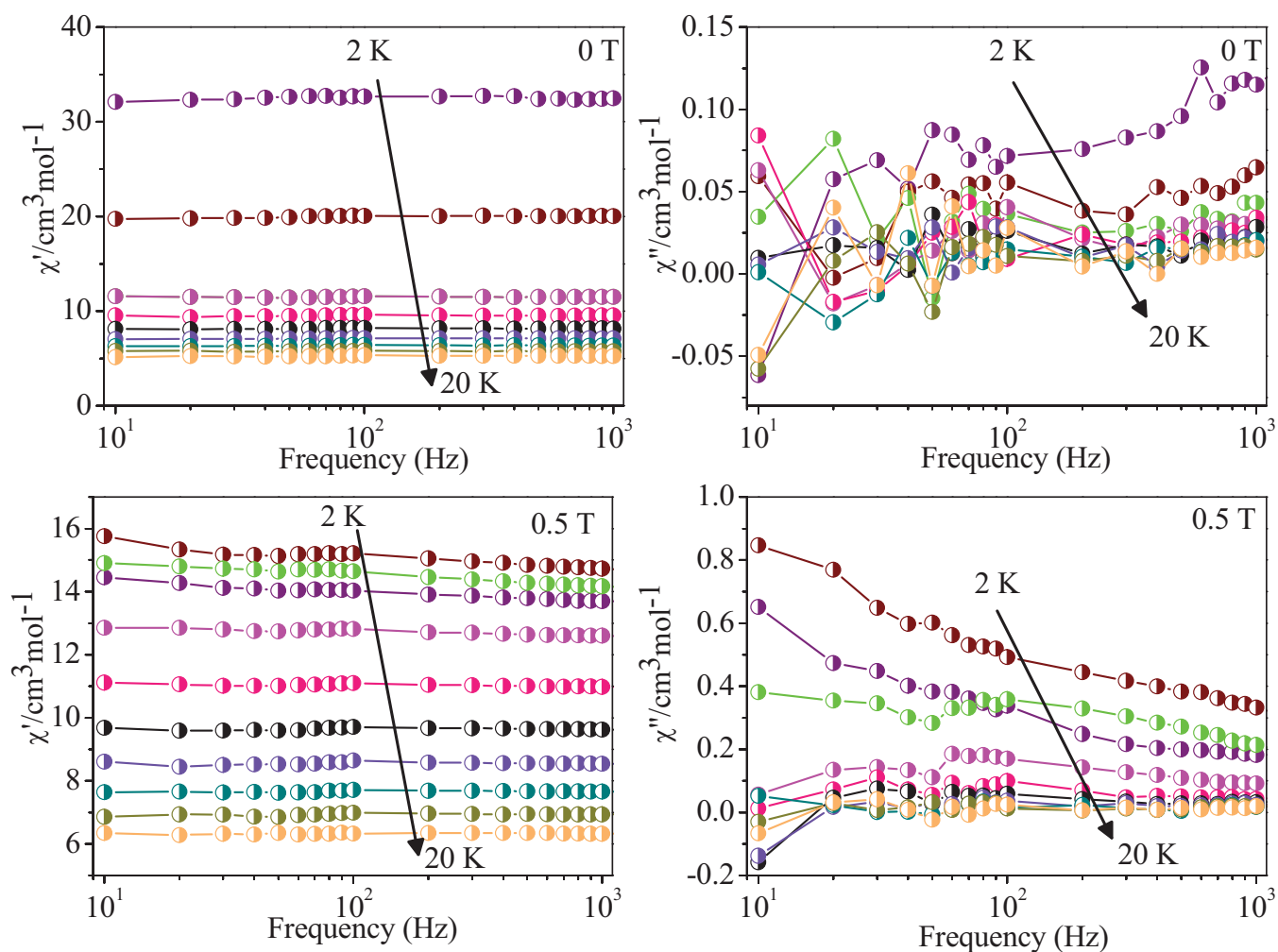


Figure S11: Frequency-dependent in-phase and out-of phase components of ac susceptibility of compound **2** under zero dc field (top) and 0.5 T field (bottom).

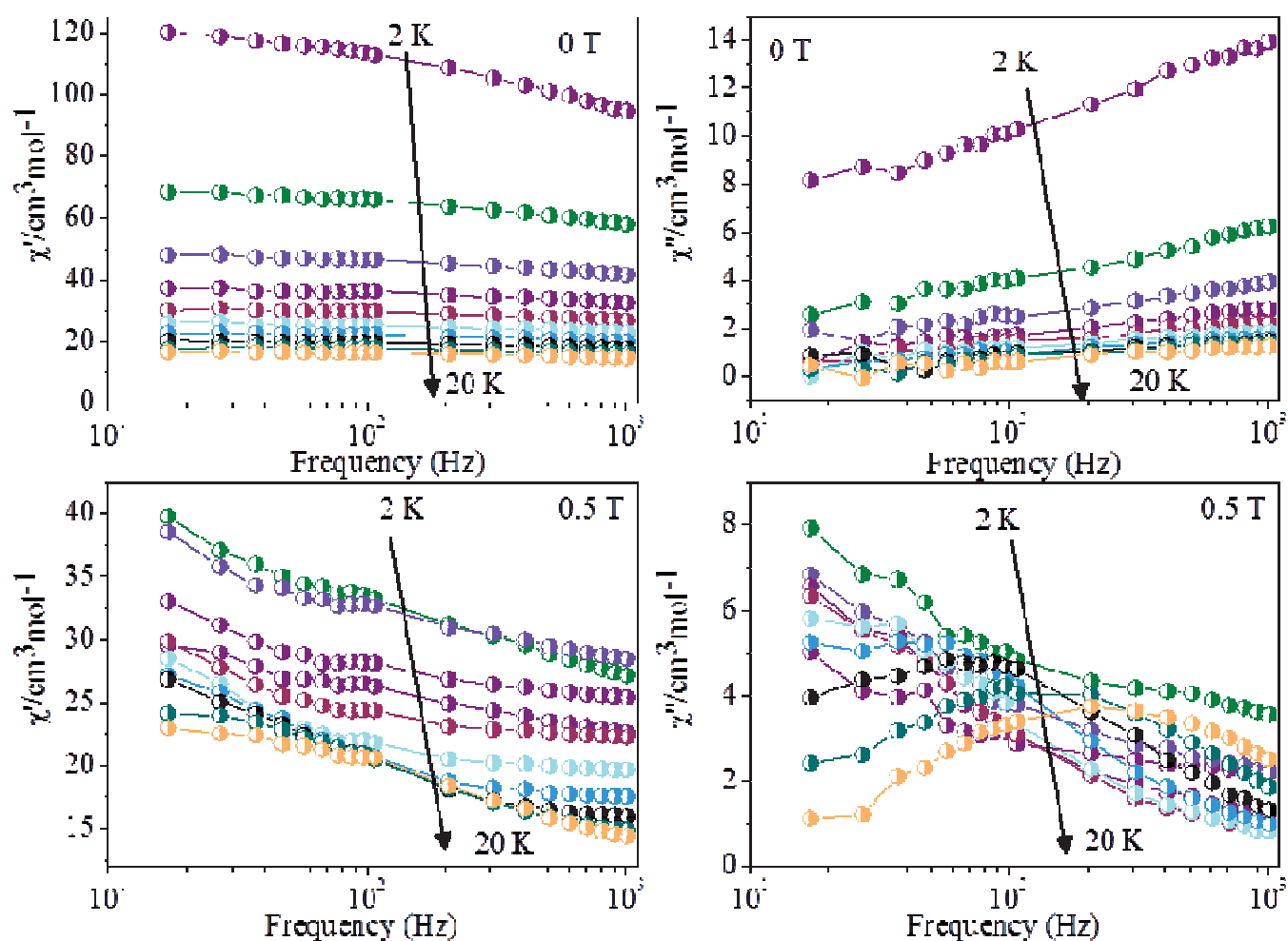


Figure S12: Frequency-dependent in-phase and out-of phase components of ac susceptibility of compound **3** under zero dc field (top) and 0.5 T field (bottom).

Table S1. Crystal data and structure refinement parameters for complexes **1-4**.

	Complex 1	Complex 2	Complex 3	Complex 4
empirical formula	C ₇₈ H ₈₈ Cl ₄ N ₂₈ O ₁₈ Dy ₈	C ₇₆ H ₈₀ Cl ₄ N ₂₈ O ₁₆ Tb ₈	C ₁₀₈ H ₁₁₄ N ₃₄ O ₂₈ Dy ₈	C ₁₂₀ H ₁₂₆ N ₃₈ O ₃₀ Nd ₆
formula weight	3147.56	3054.84	3636.33	3446.03
crystal system	Monoclinic	Monoclinic	Triclinic	Hexagonal
space group	<i>C2/c</i>	<i>C2/c</i>	<i>P-1</i>	<i>P3₂2₁</i>
<i>a</i> /Å	24.237(7)	24.275(1)	14.4489(7)	16.7861(4)
<i>b</i> /Å	19.982(6)	20.1097(10)	16.4770(9)	16.7861(4)
<i>c</i> /Å	24.195(7)	24.122(1)	18.7596(10)	47.371(2)
α /°	90	90	110.305(2)	90
β /°	101.124(7)	101.093(2)	110.042(2)	90
γ /°	90	90	94.732(2)	120
<i>V</i> /Å ³	11497(6)	11555.7(10)	3830.8(3)	11559.7(7)
reflections collected	60148	80093	52086	166758
unique reflection	11904	13954	16942	21876
observed reflections[I>2σ(I)]	10348	9295	8841	9864
<i>R</i> 1	0.0712	0.0503	0.0625	0.0856
<i>wR</i> 2	0.2037	0.1491	0.1516	0.1965
CCDC no.	1027809	1027811	1027808	1027810

Table S2. Selected bond lengths and angles of complexes 1-4

Complex 1		Dy ₂ -O ₁	2.381(6)	Dy ₃ -N _{3A}	2.441(8)	Dy ₄ -Dy ₃	3.7640(11)
Dy ₁ -O _{1S}	2.308(6)	Dy ₂ -N _{2A}	2.413(8)	Dy ₃ -O _{1A}	2.469(6)	Dy ₄ -Dy ₂	3.8205(11)
Dy ₁ -O _{2WS}	2.352(6)	Dy ₂ -N ₅	2.503(8)	Dy ₃ -O ₁	2.482(6)	Dy ₁ -O _{2WS} -Dy ₃	109.6(2)
Dy ₁ -O _{1A}	2.353(6)	Dy ₂ -N ₆	2.556(8)	Dy ₃ -Dy ₄	3.7323(11)	Dy ₁ -O _{2WS} -Dy ₄	107.0(2)
Dy ₁ -N ₂	2.470(8)	Dy ₂ -Cl ₂	2.674(2)	Dy ₃ -Dy ₄	3.7640(11)	Dy ₂ -O _{2SA} -Dy ₄	110.5(2)
Dy ₁ -N _{5A}	2.513(7)	Dy ₂ -N _{1A}	2.773(8)	Dy ₄ -O _{2S}	2.338(6)	Dy ₃ -O _{2S} -Dy ₄	108.8(2)
Dy ₁ -N _{6A}	2.591(8)	Dy ₂ -Dy ₄	3.8205(10)	Dy ₄ -O _{1SA}	2.339(6)	Dy ₁ -O _{1S} -Dy ₄	110.8(2)
Dy ₁ -Cl ₁	2.703(2)	Dy ₂ -Dy ₃	3.8612(9)	Dy ₄ -O _{2SA}	2.352(6)	Dy ₂ -O _{1WS} -Dy ₃	108.9(2)
Dy ₁ -N ₁	2.775(8)	Dy ₃ -O _{2S}	2.253(6)	Dy ₄ -O _{1S}	2.367(6)	Dy ₂ -O _{1WS} -Dy ₄	106.2(2)
Dy ₁ -Dy ₄	3.8476(9)	Dy ₃ -O _{1SA}	2.275(6)	Dy ₄ -O _{1WS}	2.426(6)	Dy ₃ -O _{1WS} -Dy ₄	102.7(2)
Dy ₁ -Dy ₃	3.8792(11)	Dy ₃ -O _{1WS}	2.394(6)	Dy ₄ -O _{2WS}	2.435(6)	Dy ₃ -O _{2WS} -Dy ₄	101.2(2)
Dy ₁ -O _{2SA}	2.299(6)	Dy ₃ -O _{2WS}	2.395(6)	Dy ₄ -N _{1S}	2.533(7)	Dy ₃ -O _{1SA} -Dy ₄	109.3(2)
Dy ₂ -O _{1WS}	2.352(6)	Dy ₃ -N ₃	2.440(7)	Dy ₄ -N _{1SA}	2.548(7)	Dy ₂ -O ₁ -Dy ₃	105.1(2)
Complex 2.		Tb ₂ -O _{1A}	2.352(6)	Tb ₃ -N _{1S}	2.531(8)	Tb ₁ -O _{2S} -Tb ₃	109.8(2)
Tb ₁ -O _{1SA}	2.251(5)	Tb ₂ -O _{1W}	2.359(5)	Tb ₃ -N _{1SA}	2.536(7)	Tb ₁ -O _{1W} -Tb ₃	101.6(2)
Tb ₁ -O _{2S}	2.267(6)	Tb ₂ -N ₂	2.465(7)	Tb ₃ -Tb ₁	3.7348(6)	Tb ₁ -O _{1SA} -Tb ₃	109.5(2)
Tb ₁ -O _{2W}	2.372(5)	Tb ₂ -N _{5A}	2.515(7)	Tb ₃ -Tb ₄	3.8251(5)	Tb ₂ -O _{2SA} -Tb ₃	111.1(2)
Tb ₁ -O _{1W}	2.398(6)	Tb ₂ -N _{6A}	2.588(8)	Tb ₃ -Tb ₂	3.8433(5)	Tb ₁ -O _{2W} -Tb ₃	103.2(2)
Tb ₁ -N _{3A}	2.411(7)	Tb ₂ -Cl ₁	2.685(3)	Tb ₄ -O _{1S}	2.292(6)	Tb ₂ -O _{1W} -Tb ₁	108.8(2)
Tb ₁ -N ₃	2.445(7)	Tb ₂ -N ₁	2.777(7)	Tb ₄ -O _{2W}	2.336(6)	Tb ₂ -O _{1W} -Tb ₃	107.1(2)
Tb ₁ -O _{1A}	2.467(5)	Tb ₂ -Tb ₃	3.8433(5)	Tb ₄ -O ₁	2.361(6)	Tb ₂ -O _{1A} -Tb ₁	106.8(2)
Tb ₁ -O ₁	2.501(5)	Tb ₃ -O _{1SA}	2.321(6)	Tb ₄ -N _{2A}	2.418(8)	Tb ₄ -O ₁ -Tb ₁	105.0(2)
Tb ₁ -Tb ₃	3.7348(6)	Tb ₃ -O _{2S}	2.328(6)	Tb ₄ -N ₅	2.508(7)	Tb ₄ -O _{1S} -Tb ₃	110.6(2)
Tb ₁ -Tb ₃	3.7585(6)	Tb ₃ -O _{1S}	2.360(6)	Tb ₄ -N ₆	2.568(8)	Tb ₄ -O _{2W} -Tb ₁	110.1(2)
Tb ₁ -Tb ₄	3.8583(6)	Tb ₃ -O _{2SA}	2.372(6)	Tb ₄ -Cl ₂	2.645(3)	Tb ₄ -O _{2W} -Tb ₃	106.9(2)
Tb ₁ -Tb ₂	3.8688(6)	Tb ₃ -O _{1W}	2.419(5)	Tb ₄ -N ₁	2.769(8)		
Tb ₂ -O _{2SA}	2.288(6)	Tb ₃ -O _{2W}	2.425(5)				
Complex 3		Dy ₂ -O _{2S}	2.343(7)	Dy ₃ O ₂	2.396(6)	Dy ₄ -N _{1B}	2.690(9)
Dy ₁ -O ₂	2.299(6)	Dy ₂ -O ₁	2.357(6)	Dy ₃ -O _{3S}	2.448(9)	Dy ₁ -O _{6S} -Dy ₁	111.4(2)
Dy ₁ -O ₁	2.356(6)	Dy ₂ -O _{1S}	2.360(6)	Dy ₃ -O _{1S}	2.470(6)	Dy ₁ -O _{6S} -Dy ₃	106.7(2)
Dy ₁ -O _{3S'}	2.376(11)	Dy ₂ -O _{1B}	2.361(6)	Dy ₃ -N _{5B}	2.501(8)	Dy ₁ -O ₁ -Dy ₂	114.0(3)
Dy ₁ -O _{1S}	2.392(7)	Dy ₂ -N ₃	2.386(8)	Dy ₃ -Dy ₁	3.8424(6)	Dy ₁ -O _{1S} -Dy ₃	99.9(2)
Dy ₁ -O _{6S}	2.393(6)	Dy ₂ -N _{3B}	2.404(7)	Dy ₄ -O _{2A}	2.221(8)	Dy ₁ -O ₂ -Dy ₃	109.8(3)
Dy ₁ -O _{6S}	2.406(6)	Dy ₂ -N _{3A}	2.454(8)	Dy ₄ -O _{1D}	2.421(8)	Dy ₂ -O _{1S} -Dy ₁	112.6(3)
Dy ₁ -O _{4S}	2.464(7)	Dy ₂ -O _{1A}	2.491(6)	Dy ₄ -N _{2B}	2.439(9)	Dy ₂ -O _{1S} -Dy ₃	108.8(2)
Dy ₁ -N ₅	2.519(8)	Dy ₂ -Dy ₃	3.9271(7)	Dy ₄ -O _{1A}	2.445(6)	Dy ₃ -O _{6S} -Dy ₁	101.6(2)
Dy ₁ -Dy ₃	3.7217(7)	Dy ₃ -O _{2B}	2.265(7)	Dy ₄ -N ₂	2.462(8)	Dy ₃ -O _{1B} -Dy ₂	113.2(2)
Dy ₁ -Dy ₃	3.8424(6)	Dy ₃ -O _{7S}	2.311(7)	Dy ₄ -O _{1DA}	2.467(10)	Dy ₄ -O _{1A} -Dy ₂	123.2(3)
Dy ₁ -Dy ₂	3.9532(7)	Dy ₃ -O _{1B}	2.343(6)	Dy ₄ -N _{5A}	2.496(10)		
Dy ₁ -Dy ₁	3.9657(10)	Dy ₃ -O _{6S}	2.395(6)	Dy ₄ -N ₁	2.619(9)		

Complex 4

Nd₁-O_{1D} 2.420(19)
Nd₁-O_{1A} 2.424(8)
Nd₁-O_{1A} 2.424(8)
Nd₁-O_{2A} 2.455(9)
Nd₁-O_{2A} 2.455(9)
Nd₁-O_{1W} 2.528(10)
Nd₁-O_{1W} 2.528(10)
Nd₁-N_{5A} 2.592(12)
Nd₁-N_{5A} 2.592(12)

Nd₂-O_{1B}-Nd₃ 110.1(3)

Nd₃-O_{1W}-Nd₁ 96.9(3)

Nd₄-O₁-Nd₁ 121.6(3)

Nd₁-Nd₃ 3.7390(9)

Nd₂-O_{2B} 2.353(10)

Nd₂-O_{2B} 2.353(10)

Nd₂-O_{1W} 2.458(9)

Nd₂-O_{1W} 2.458(9)

Nd₂-O_{1B} 2.513(8)

Nd₂-O_{1B} 2.513(8)

Nd₂-N_{5B} 2.661(12)

Nd₂-N_{5B} 2.661(12)

Nd₃-O_{1W} 2.469(7)

Nd₃-O_{2A} 2.484(10)

Nd₃-O_{1A} 2.491(9)

Nd₃-O_{1WS} 2.505(10)

Nd₃-N_{3A} 2.534(11)

Nd₃-N_{3B} 2.549(12)

Nd₃-O_{1B} 2.576(9)

Nd₃-O₁ 2.609(8)

Nd₃-N₃ 2.615(11)

Nd₄-O₂ 2.297(9)

Nd₄-O_{4N} 2.405(12)

Nd₄-O₁ 2.513(8)

Nd₄-N_{2A} 2.515(11)

Nd₄-O_{1SS} 2.54(2)

Nd₄-N_{2B} 2.543(13)

Nd₄-N₅ 2.598(12)

Nd₄-N_{1A} 2.71(1)

Nd₄-N_{1B} 2.760(13)

Nd₁-O_{2A}-Nd₃ 130.8(8)

Nd₁-O_{1A}-Nd₃ 99.1(3)

Nd₂-O_{1W}-Nd₁ 110.1(3)

Nd₂-O_{1W}-Nd₃ 115.7(3)

References:

S1. (a) Alexandropoulos, D. I; Mukherjee, S; Papatriantafyllopoulou, C; Raptopoulou, C. P; Psycharis, V; Bekiari, V; Christou, G; Stamatatos, T. C. *Inorg. Chem.***2011**, 50, 11276–11278.(b) Canaj, A. B; Tzimopoulos, D. I; Philippidis, A; Kostakis, G. E; Milios, C. *J. Inorg. Chem.* **2012**, 51, 7451-7453.